

Reinhold Environmental Ltd.



2009 NOx-Combustion Round
Table & Expo Presentation

February 9 & 10, 2009, Cleveland, OH

INTEGRATED CONDENSABLES MANAGEMENT

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Condensables Measurement System

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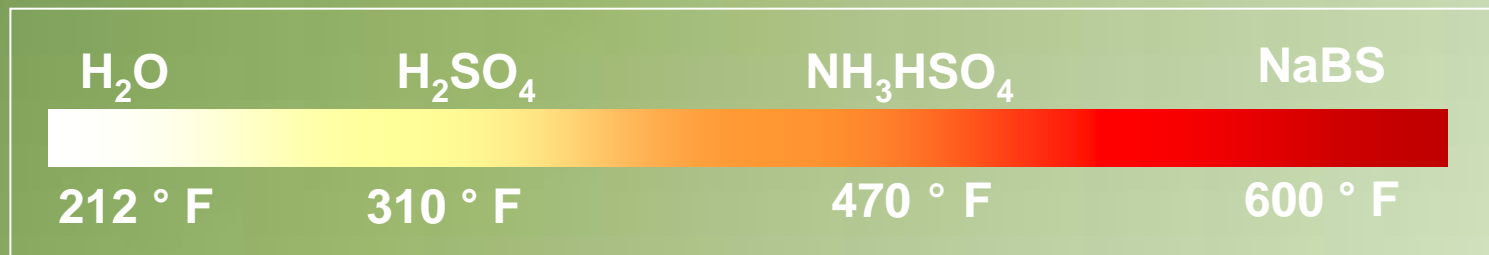
AbSensor – AbS/SO₃

What is it? What does it do?

- In-Situ, Continuous measurement
- Temp at which material condenses out from flue gas

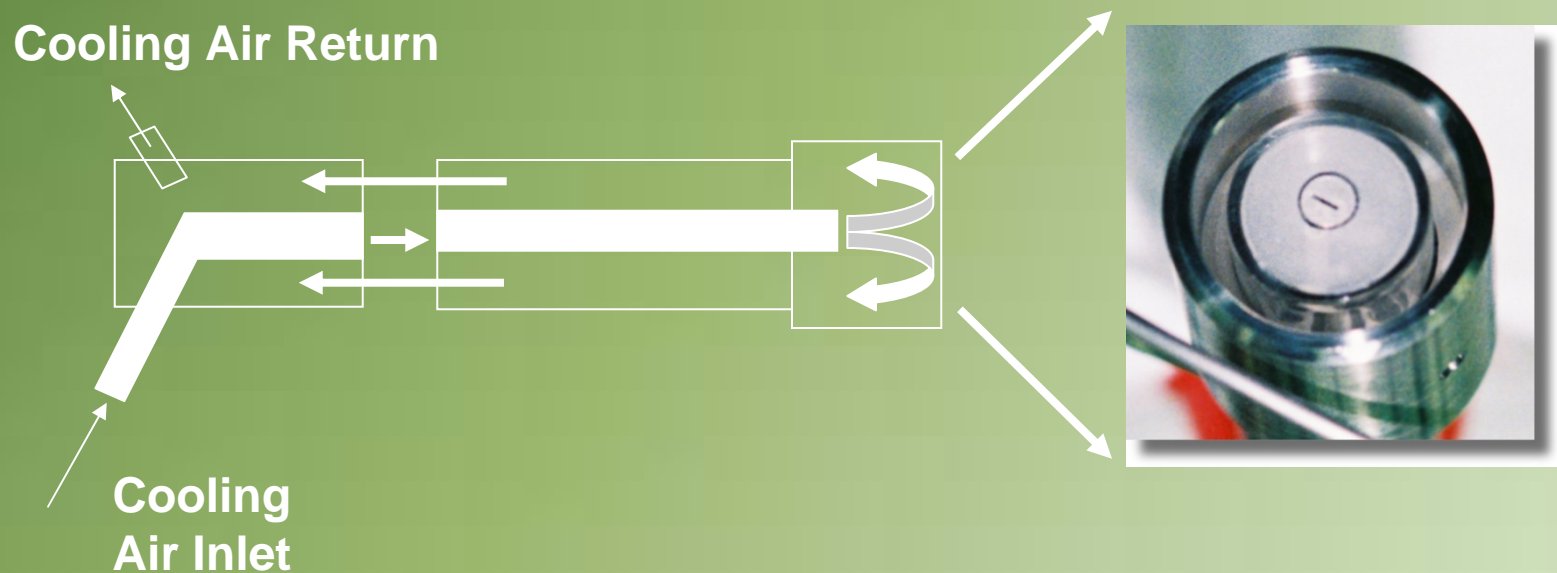
This material could be:

- Moisture (H₂O),
- Sulfuric Acid (H₂SO₄) (H₂O + SO₃)
- Ammonium Bisulfate (NH₃HSO₄) (NH₃ + H₂O + SO₃)
- Sodium Bisulfate



The same device measures condensables across the spectrum!

How does it work? - I



Cooling air flow to the probe tip is precisely controlled to induce condensation on the probe surface

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AbSensor - Introduction

Condensation = Evaporation

Eq

Tip Temp

The Condensables System Output is:

Formation Temp: The temperature at which material will first form

The Equilibrium Dewpoint

Evaporation Temp: The temperature at which material will self-evaporate

Tip Current

A hot probe is precisely cooled until condensation current is detected. (Formation Point).

Condensation > Evaporation

The probe is allowed to heat in the Flue Gas until the current goes below a threshold (Evaporation Point).

Evaporation > Condensation

Cooling Air Return

Cooling Air Inlet

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AbSensor – AbS/SO3 System



- 4" 150 lb 8-bolt flanged port
- 50 psi service air
- 12 cfm air consumption
- 110 VAC power supply

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MOT

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What is SCR Minimum Operating Temperature (MOT)?

- ⦿ This is the recommended temperature below which there is a concern of catalyst plugging due to condensation of Ammonium Bisulfate in the catalyst pores leading to deactivation of the catalyst and reduced SCR performance
- ⦿ Below this temperature the NH_3 injection is turned off

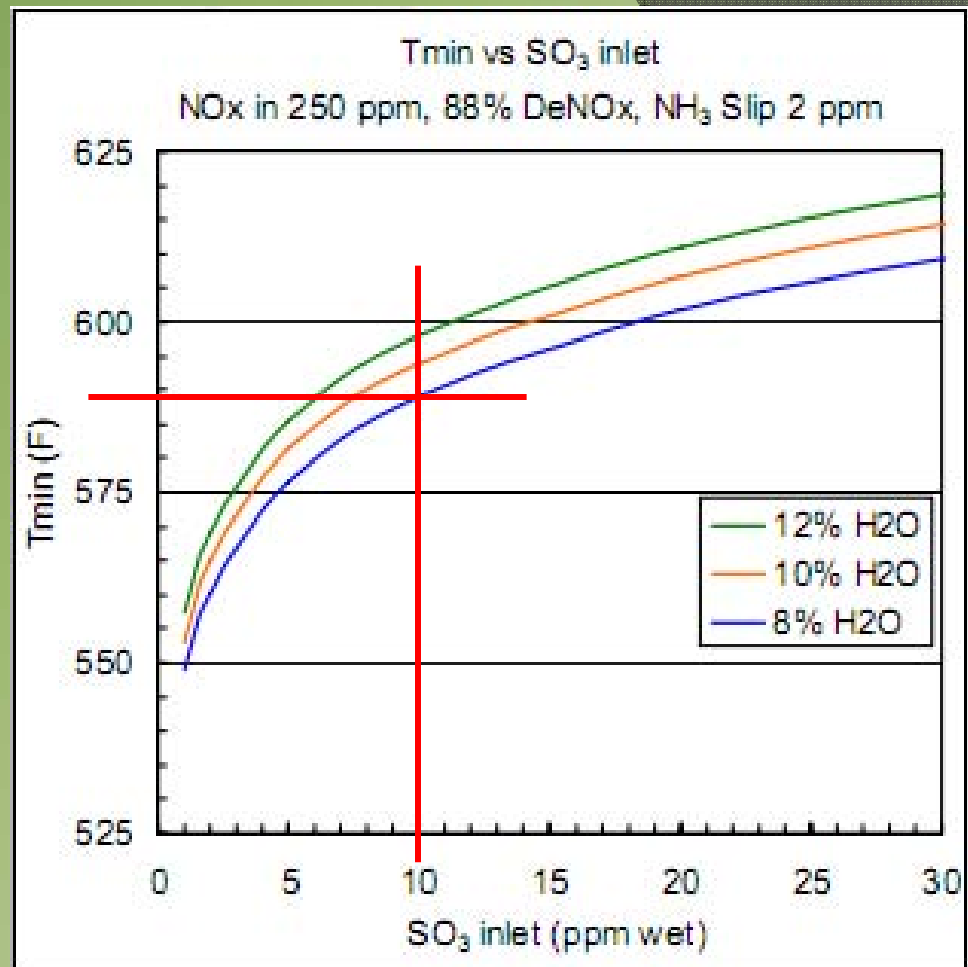
What is the AbS Dewpoint?

- AbS Dewpoint_{BULK} =
func(SO₃, NH₃, Moisture)
- AbS Dewpoint_{CAPILLARY} =
func(AbS Dewpoint_{BULK}, Pore Size, Activity?)
- Typically in the range of 330 to 550 DegF, but
could be higher

Vendor MOT Curves

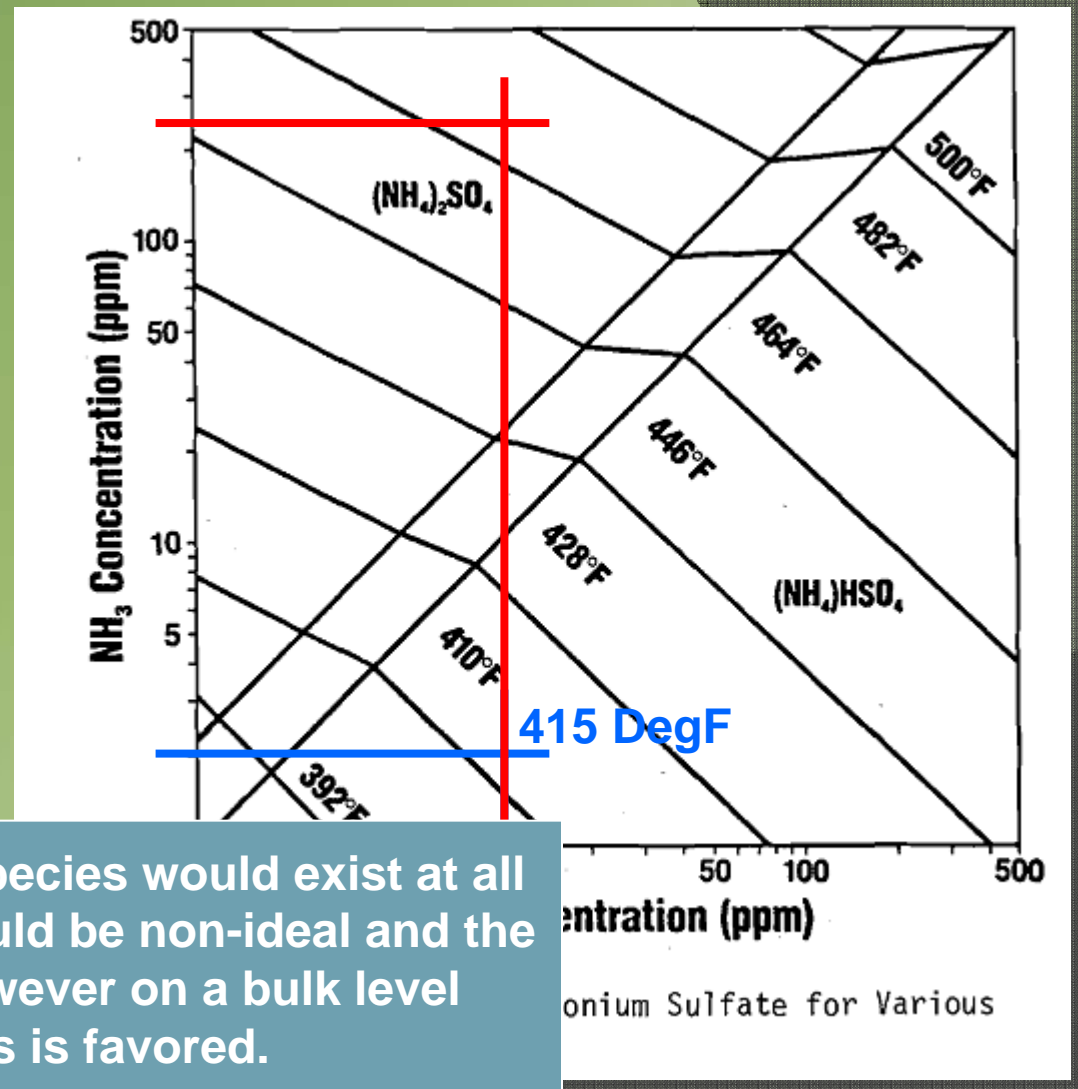
- Typical catalyst vendor curve showing minimum operating temperature based on a 2 ppm NH₃ slip at various SO₃ and moisture levels.
- A typical case of 10 ppm SO₃ at 8 % moisture yields a MOT of 595 DegF
- These curves take into account the pore size of the specific catalyst design

Source: Argillon Presentation at the Reinhold Environmental NO_x Round Table 2008



AbS – AS Phase Diagram Hitachi-Zosen Paper

- 2 ppm NH₃ Slip, 10 ppm SO₃ = Formation Temp of 415 DegF with potential for AbS
- At SCR inlet the NH₃ would be 250 ppm with 10 ppm SO₃ = Formation Temp of 485 DegF and potential for AS



In reality, it is expected that both species would exist at all times as the spatial distribution would be non-ideal and the equilibria could shift locally. However on a bulk level one or the other species is favored.

AbS Dewpoint Lab Tests by Matsuda et. al.

- The apparatus shown here was used by Matsuda et. al. to determine the formation of AS v/s AbS and the dewpoint of AbS in 1982 using a Quartz Tube Reactor

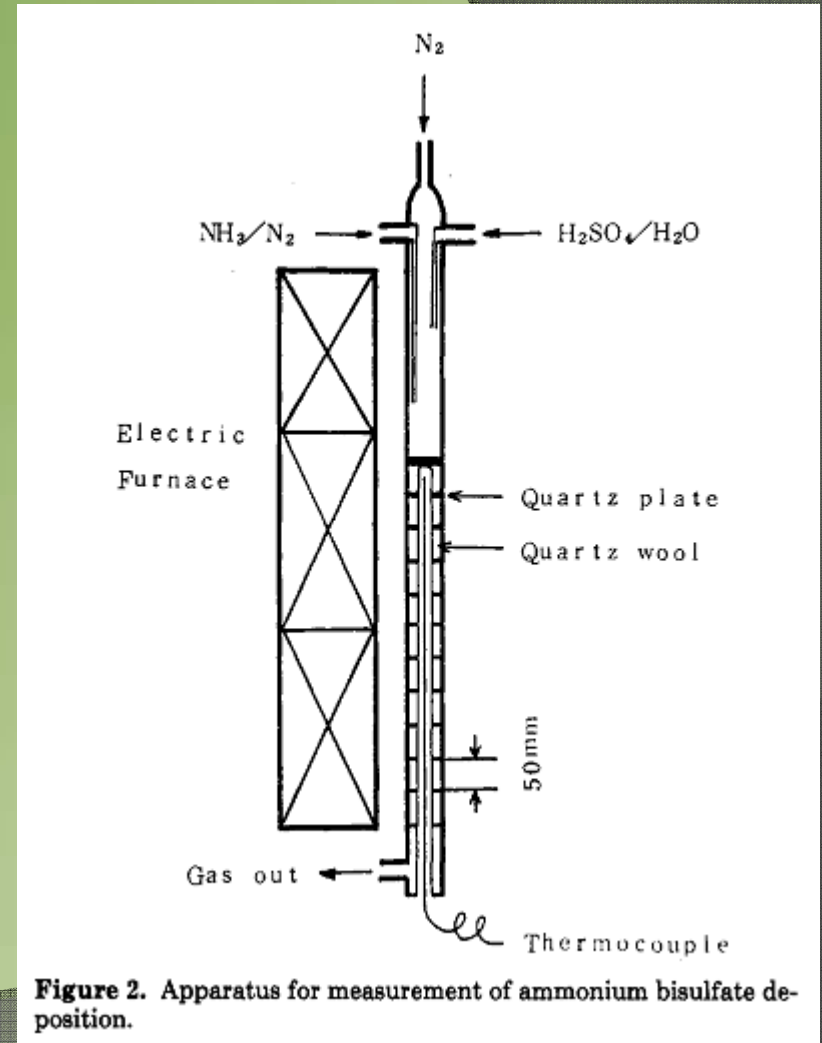
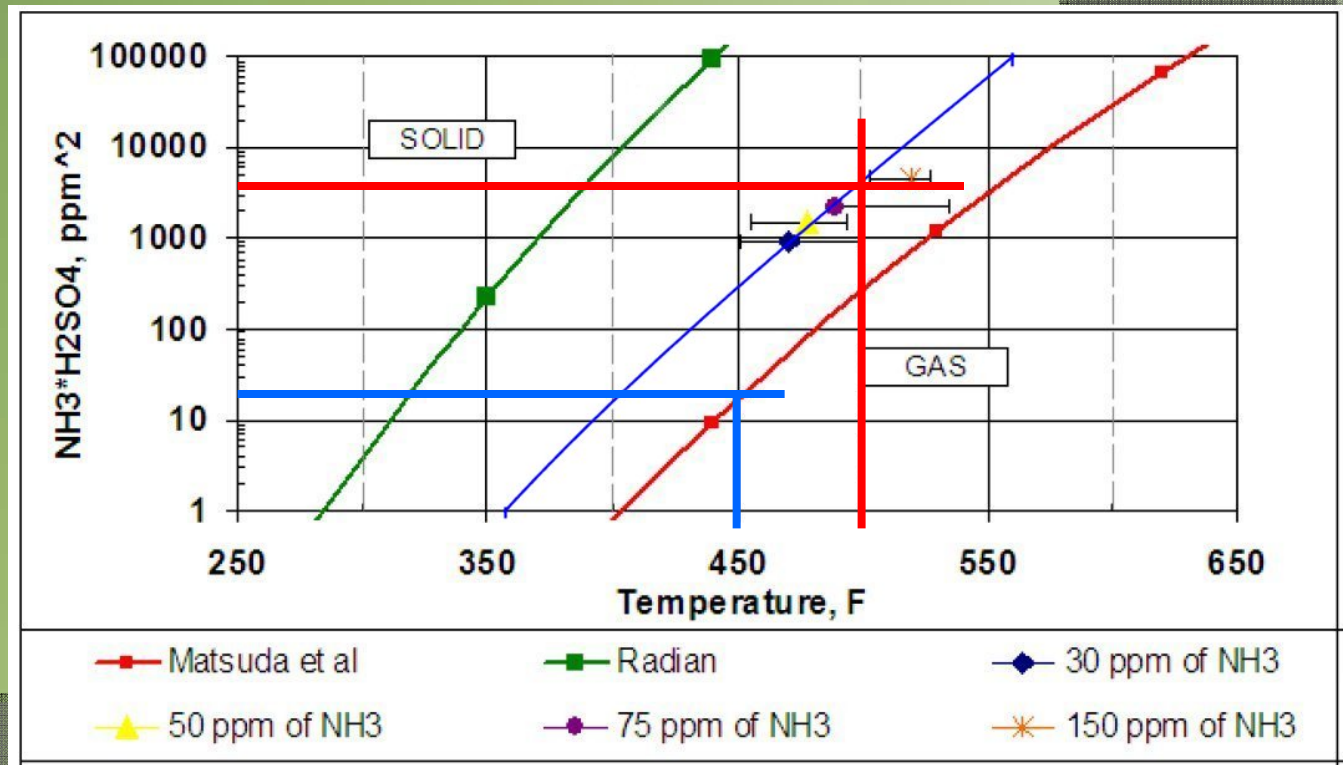


Figure 2. Apparatus for measurement of ammonium bisulfate deposition.



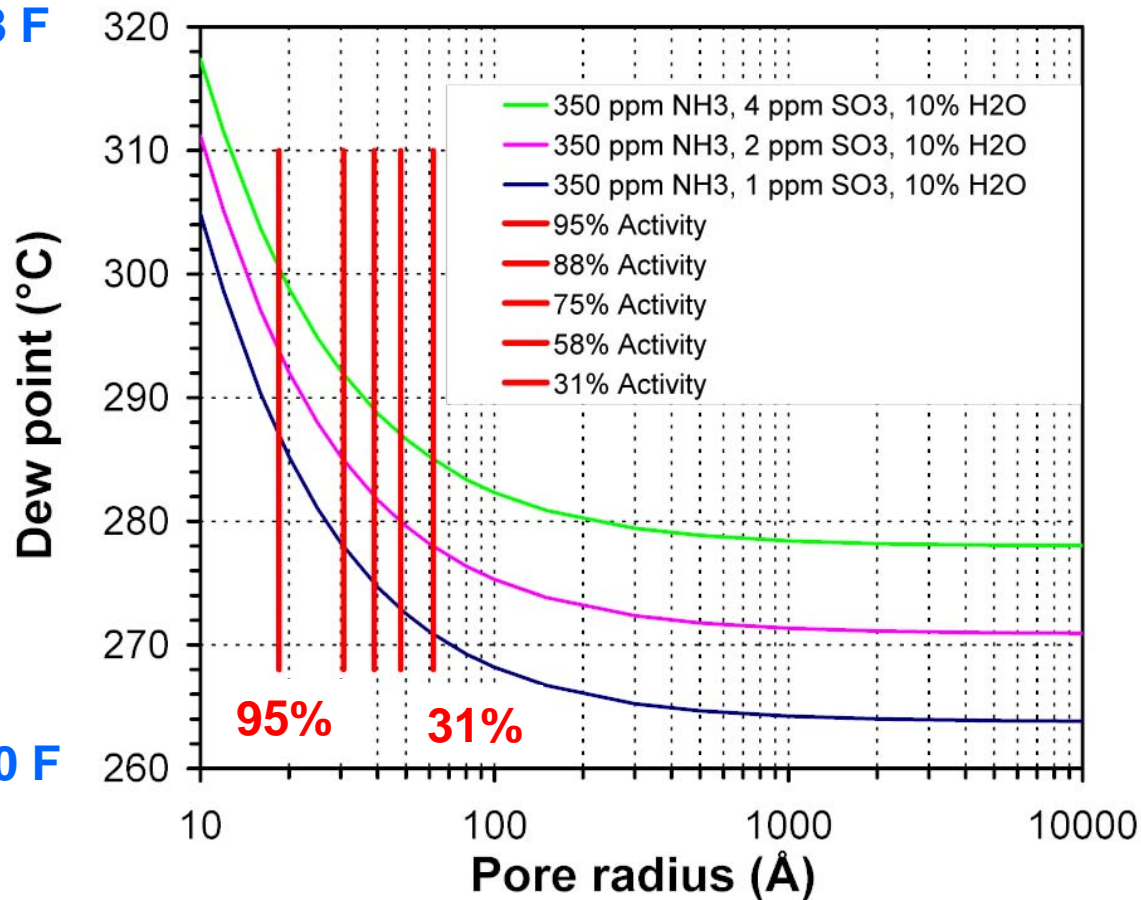
- UCI work in conjunction with FERCO, EPRI and Breen Energy produced the graph below showing AbS dewpoint as a function of NH_3 and SO_3 ppm. The data varies widely between Radian, Matsuda and those collected under this program

- 2 ppm NH_3 * 10 ppm $\text{SO}_3 = 20$ ppm² $\text{NH}_3 * \text{SO}_3$
- AbS Dewpoint of 450 DegF
- SCR inlet 250 ppm NH_3 * 10 ppm $\text{SO}_3 = 2500$ ppm² $\text{NH}_3 * \text{SO}_3$
- AbS Dewpoint of 500 DegF



Capillary Dewpoint = Func(SO3, Pore Size, Activity)

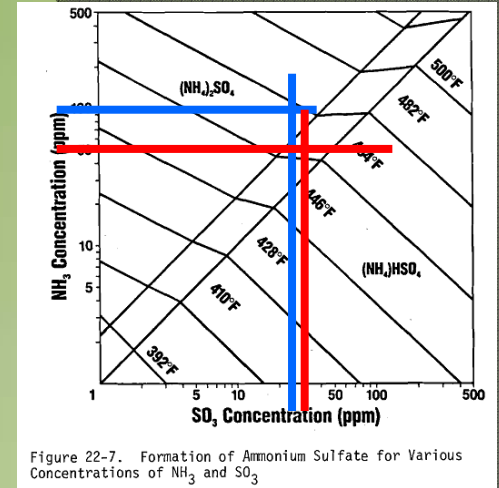
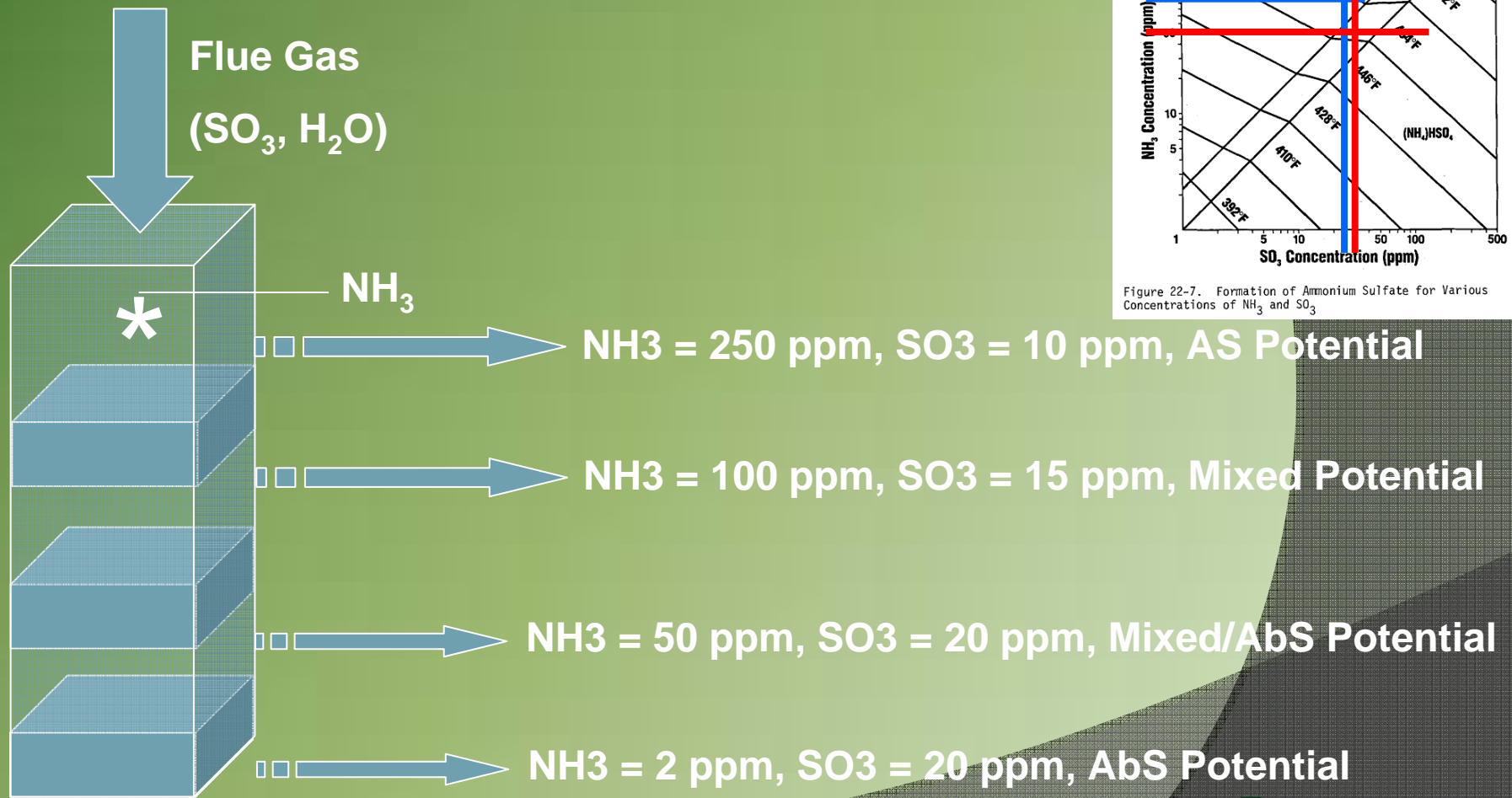
608 F



The effective pore size increases as Catalyst Activity decreases

500 F

Equilibrium within the SCR



Have I confused you enough
by now?

Can you measure AbS Dewpoint?

- Yes! AbSensor – AbS can measure the bulk dewpoint of Ammonium Bisulfate

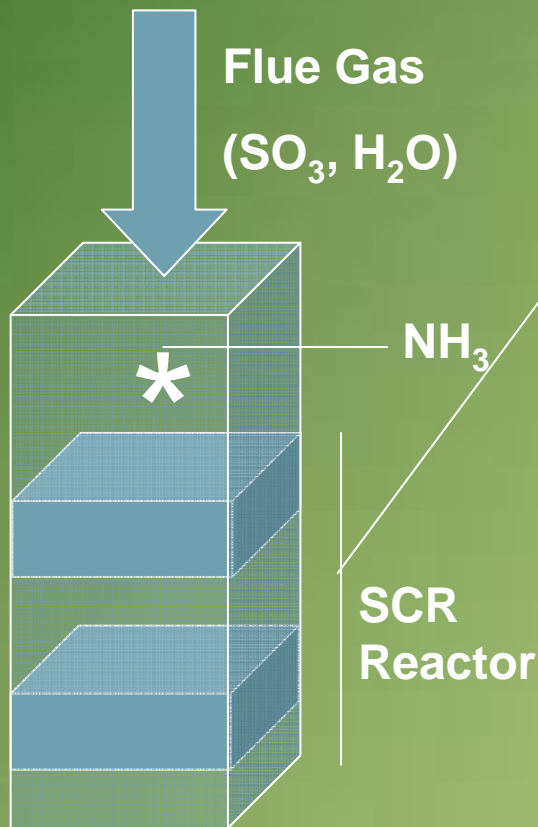
Why lower MOT?

- ⦿ Allow the SCR to continue operating at lower loads:
 - Higher Load turn-down – AGC, Wind Energy
 - Regulatory compliance for SCR operation
 - Tradable NO_x Credits
- ⦿ Heat Rate improvement, if the temperatures are held higher artificially using Air Bypass or Heating coils

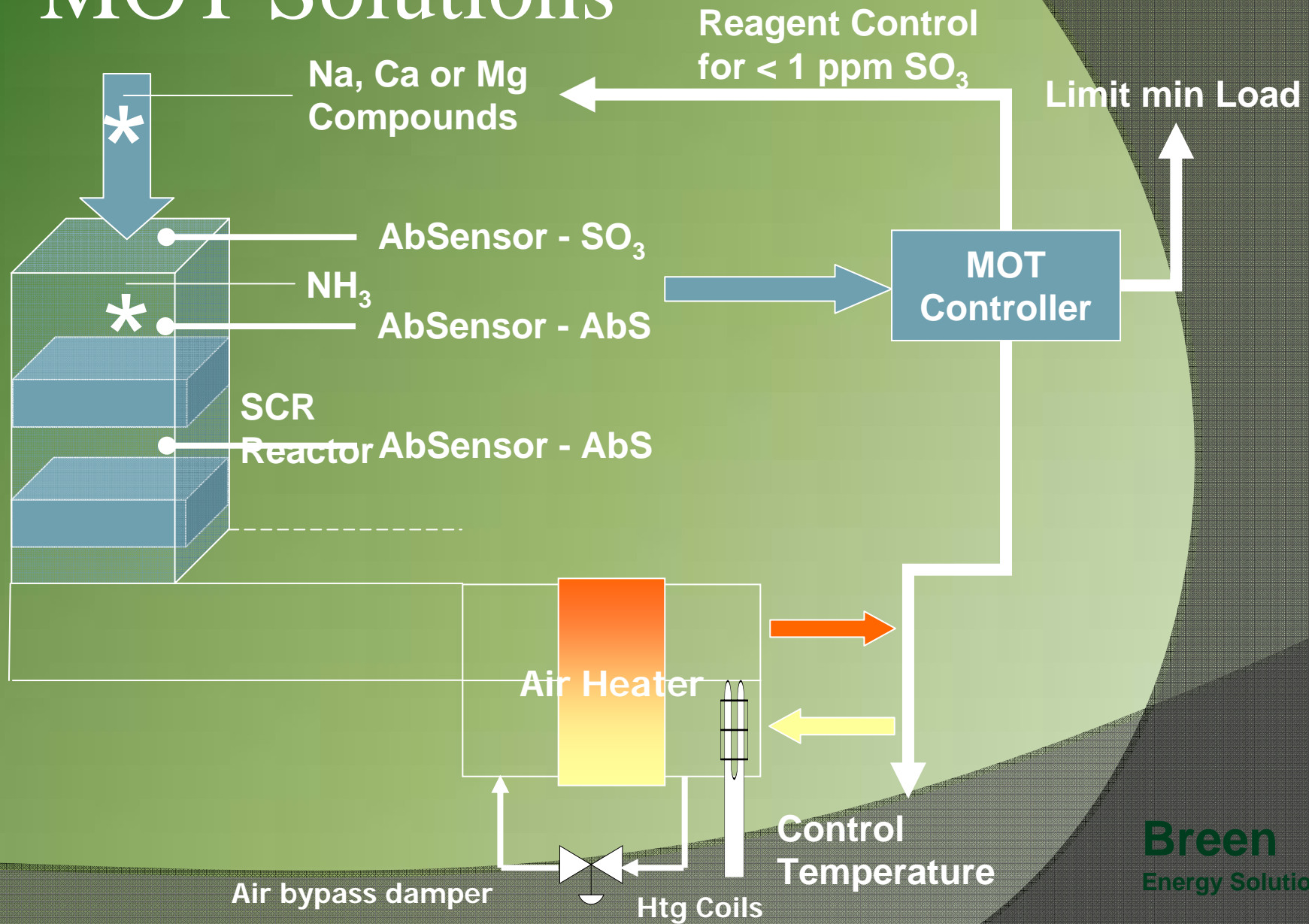
How can you lower MOT?

- Burn Low-S Coal
- Lower the SO_3 coming into the SCR using chemical reagents such as Na, Ca or Mg compounds
- Measure the SO_3 coming into the SCR and the AbS dewpoint post- NH_3 injection prior to the SCR and run the SCR just above the AbS dewpoint

SCR Operation - Reactions



MOT Solutions



Ideal solution

- Reduce the SO_3 coming into the SCR down to 1 ppm thereby shifting the equilibrium heavily towards AS throughout the SCR

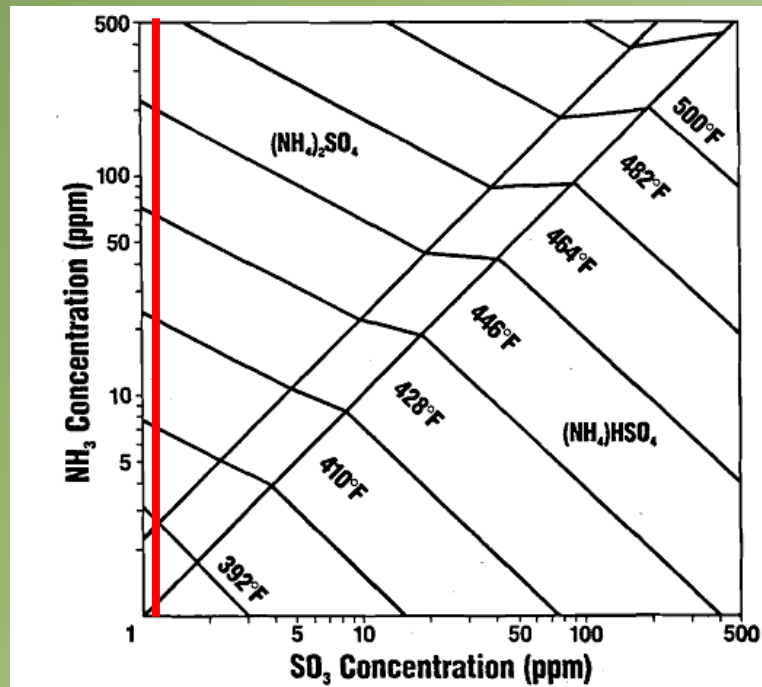
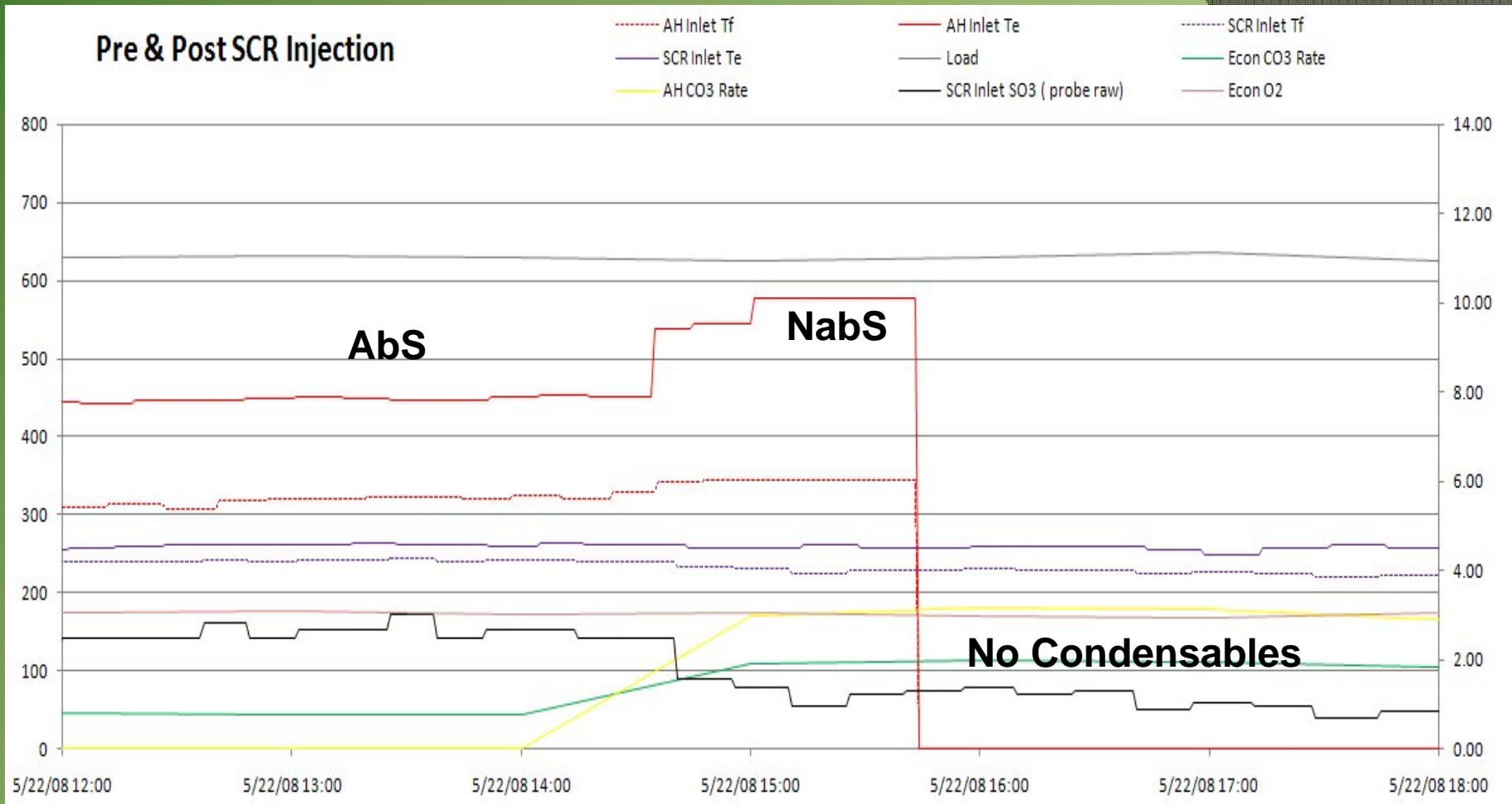


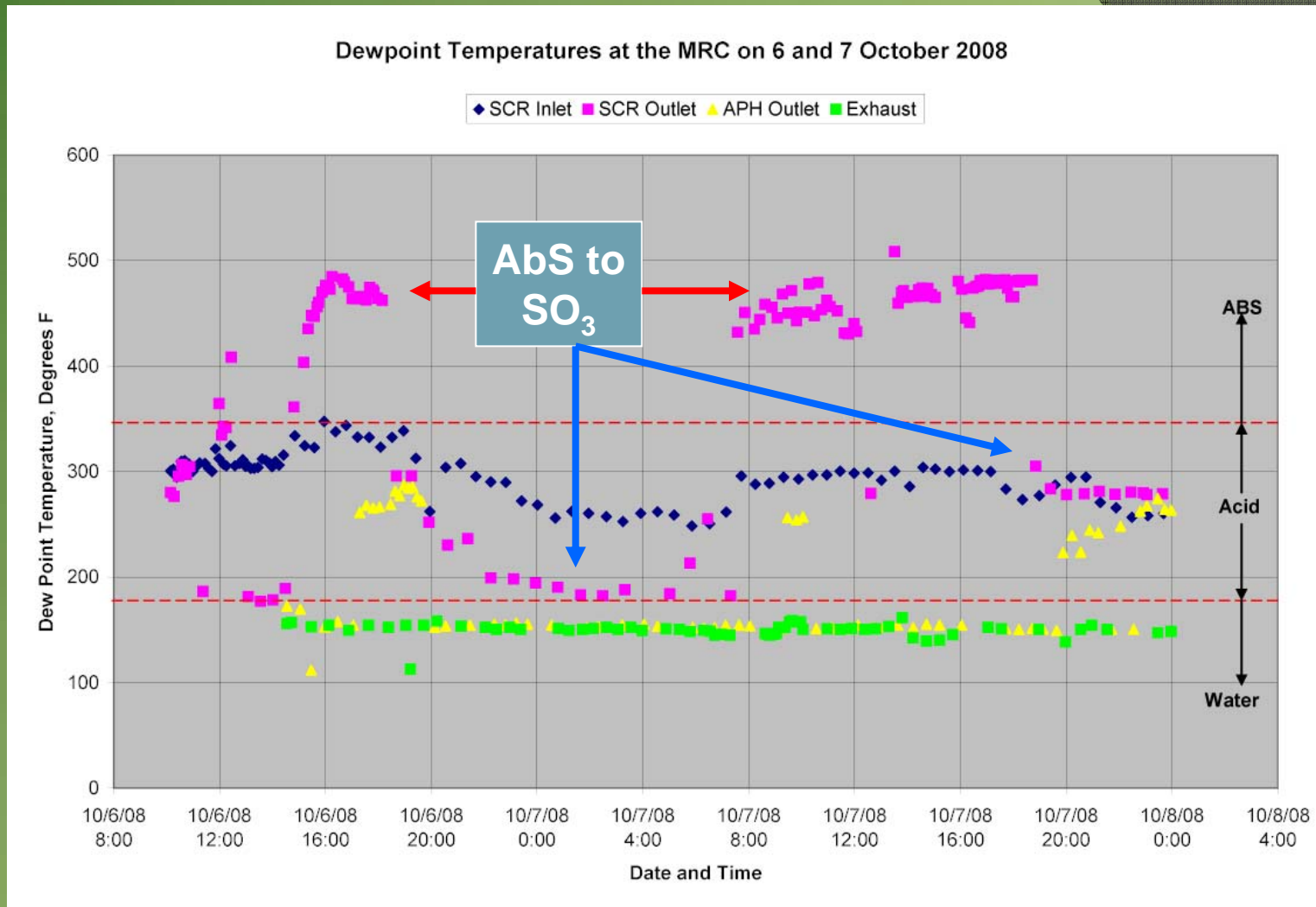
Figure 22-7. Formation of Ammonium Sulfate for Various Concentrations of NH_3 and SO_3

SCR Inlet SO₃ control – Na* Example



* URS SBS Injection System

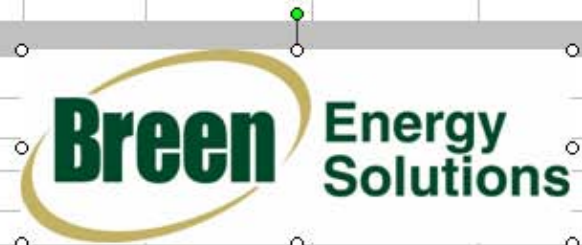
SCR Inlet SO₃ control – Ca* Example



* Test conducted at Mercury Research Center at Southern Company Plant Crist with Chemical Lime

Breen Tool for SO₃

	A	B	C	D	E	F	G	H	I	J	K	L
1												
2			Common						Factors			
3			Duct Pressure=	-10	"H2O				A	2.276		
4			Duct Pressure=	741.3176008	mmHg				B	0.0294		
5			P(H2O)=	10.0%	mol %				C	0.0858		
6			P(H2O)=	74.13176008	mmHg				D	0.0062		
7												
8	-0.130932		SO3 to DP									
9	2.570214		Given:					Min (>0)	Max			
10	2.439282		SO3 =	10	ppm		Range:	1	100		Plot Curves	
11	136.8066		Calculate:									
12	278.2519		Calc-Tdp =	278	°F							
13												
14	4.70142		DP to SO3									
15	3485.245		Given:									
16	-5.659217		Tdp =	265	°F							
17	0.334481		Calculate:									
18			SO3 =	4.7	ppm		7					
19												
20												
21			Measured DP	Measure SO3	Dewpoint to SO3 ppm			Factors		Tune:	10%	
22			250	5	And			A	2.276	2.5036	2.048	
23		260	7	B				0.0294	0.03234	0.0264		
24		270	9	C				0.0858	0.09438	0.0772		
25		280	17	D				0.0062	0.00682	0.0055		
26		290	25	SO3 ppm to Dewpoint								
27												
28												



Breen Tool for AbS

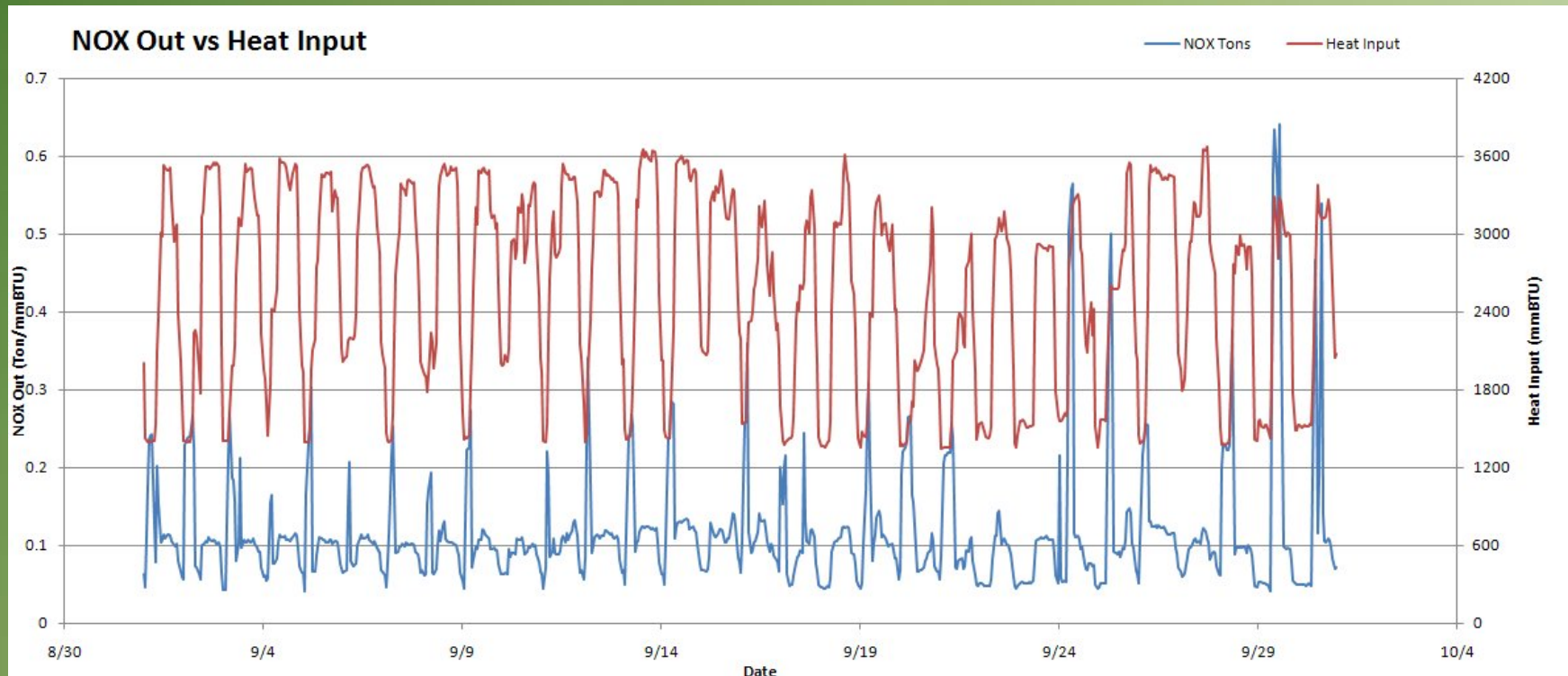
AbS Dewpoint = f(NH3, SO3)		SO3 = f(AbS Dewpoint, NH3)		NH3 = f(AbS Dewpoint, SO3)	
Given:		Given:		Given:	
NH3	1 ppm	NH3	1 ppm	SO3	3 ppm
SO3	3 ppm	AbS Dewpoint	424 DegF	AbS Dewpoint	424 DegF
		AbS Dewpoint	490.927778 DegK	AbS Dewpoint	490.9278 DegK
		Flue Gas Temp		Flue Gas Temp	
Intermediate:		Intermediate:		Intermediate:	
P(NH3)	0.000001 atm	P(NH3)	0.000001 atm	P(SO3)	0.000003 atm
P(SO3)	0.000003 atm				
Calculate Bulk Dewpoint:		Calculate:		Calculate:	
AbS Dewpoint:	491.2725503 DegK	SO3	2.887762439 ppm	NH3	0.962587 ppm
AbS Dewpoint:	424.6205906 DegF				
Calculate Capillary Dewpoint:					
Pore Radius:	25 Angstrom				
Pore Activity:	0.150000908				
Capillary AbS Dewpoint:	509.0597113 DegK				
Capillary AbS Dewpoint:	456.6374804 DegF				

- AbS Bulk Dewpoint from NH₃ and SO₃
- AbS Capillary Dewpoint from Bulk Dewpoint and Pore Size
- SO₃ ppm from AbS Bulk Dewpoint and NH₃ Slip
- NH₃ slip from AbS Bulk Dewpoint and SO₃ ppm



By: Chetan Chothani

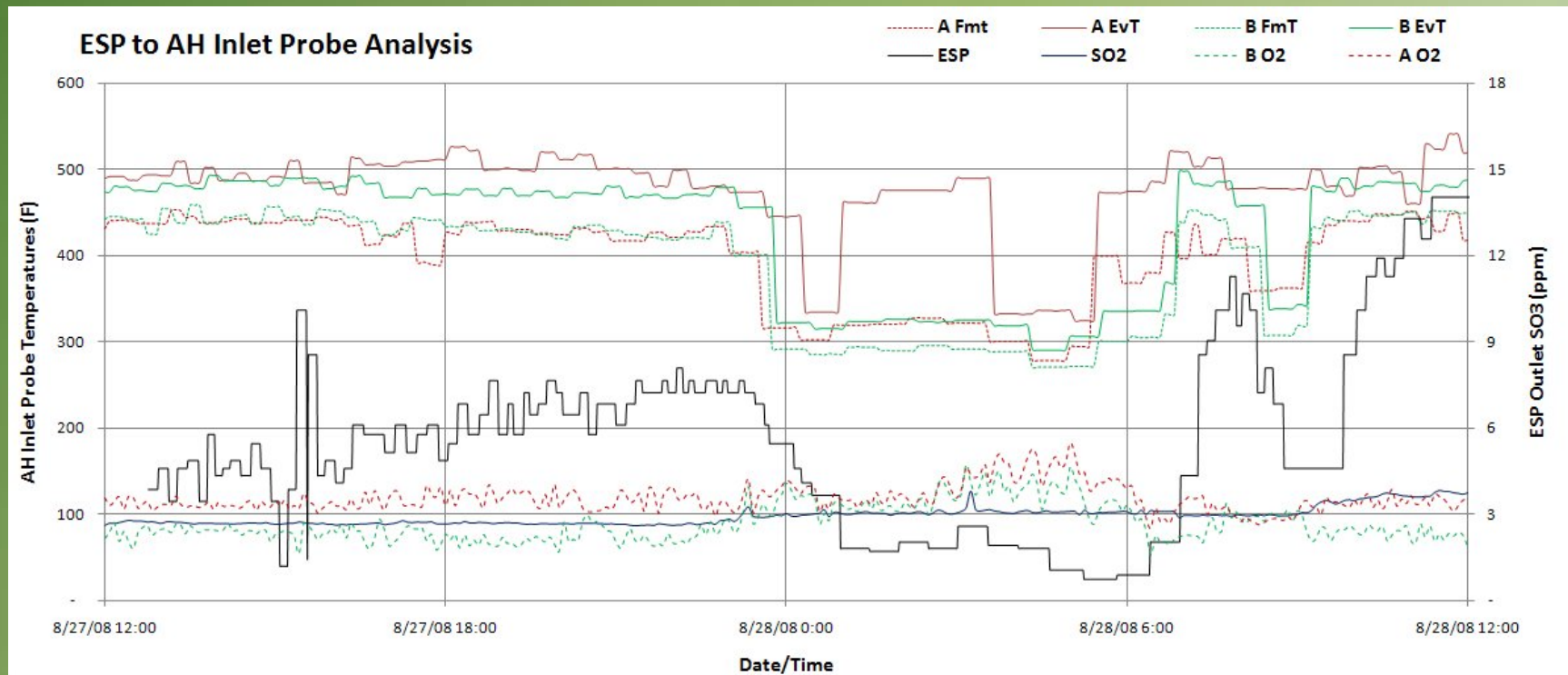
NOX Out vs. Heat Input



Reduction of overnight NOX spikes can exceed 250 Tons/Year on a 600MW Plant!!

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Morning Sickness Surge



Dynamic Air Heater Cleaning

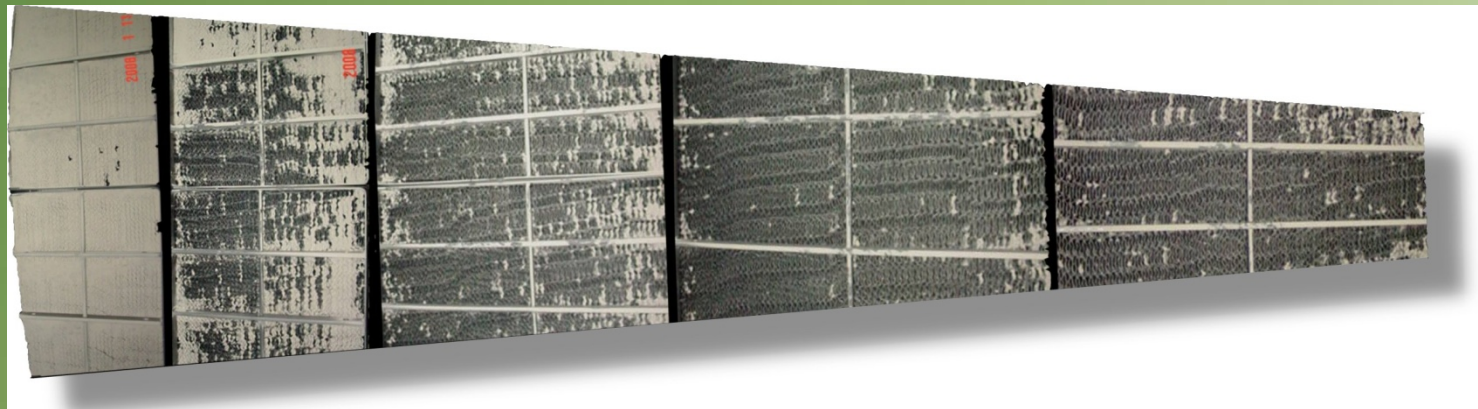
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Air Heater Sootblowing



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Fouling Distance from Hub

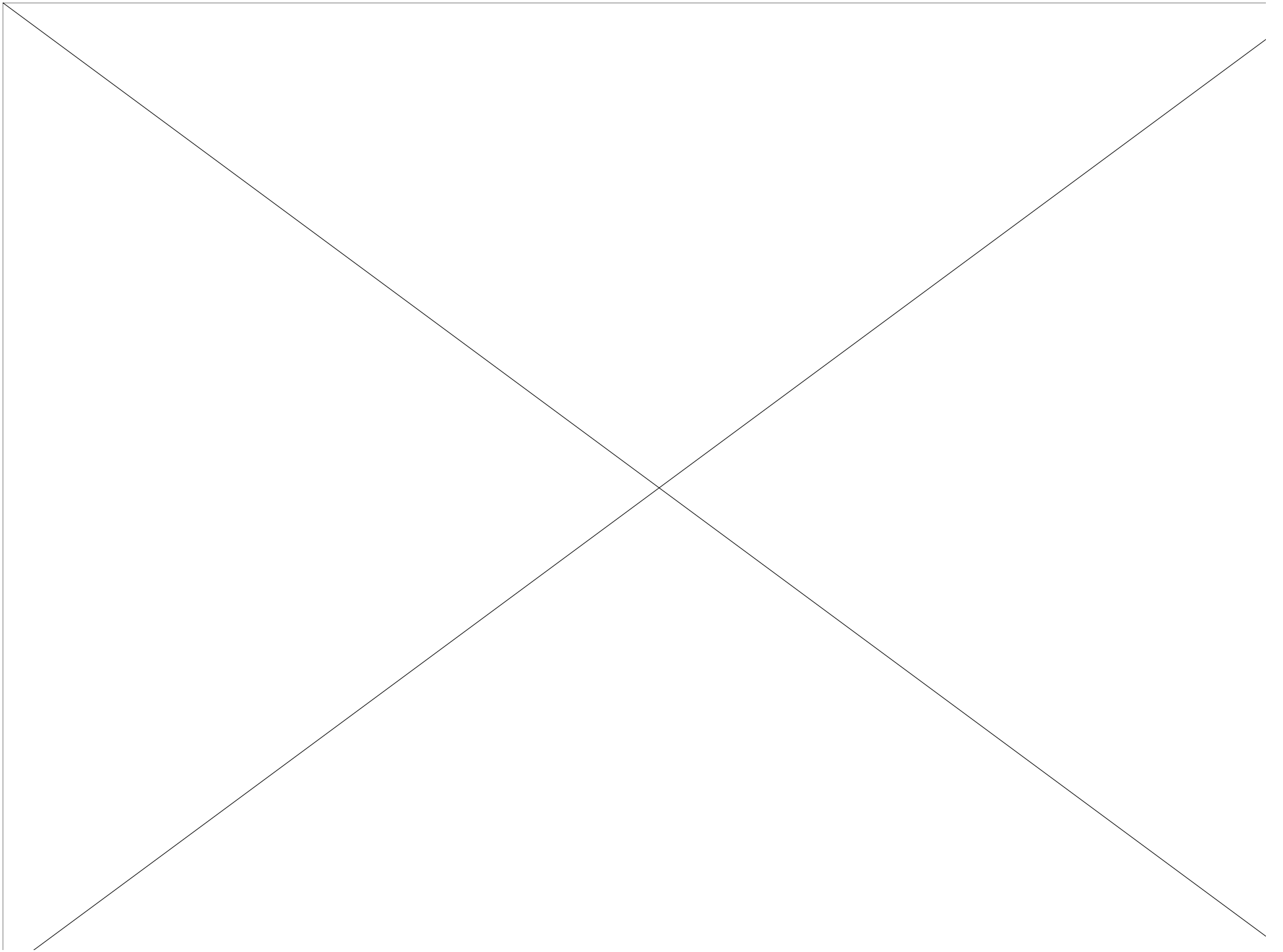


Outermost
Baskets



Increasing Fouling

Innermost
Baskets

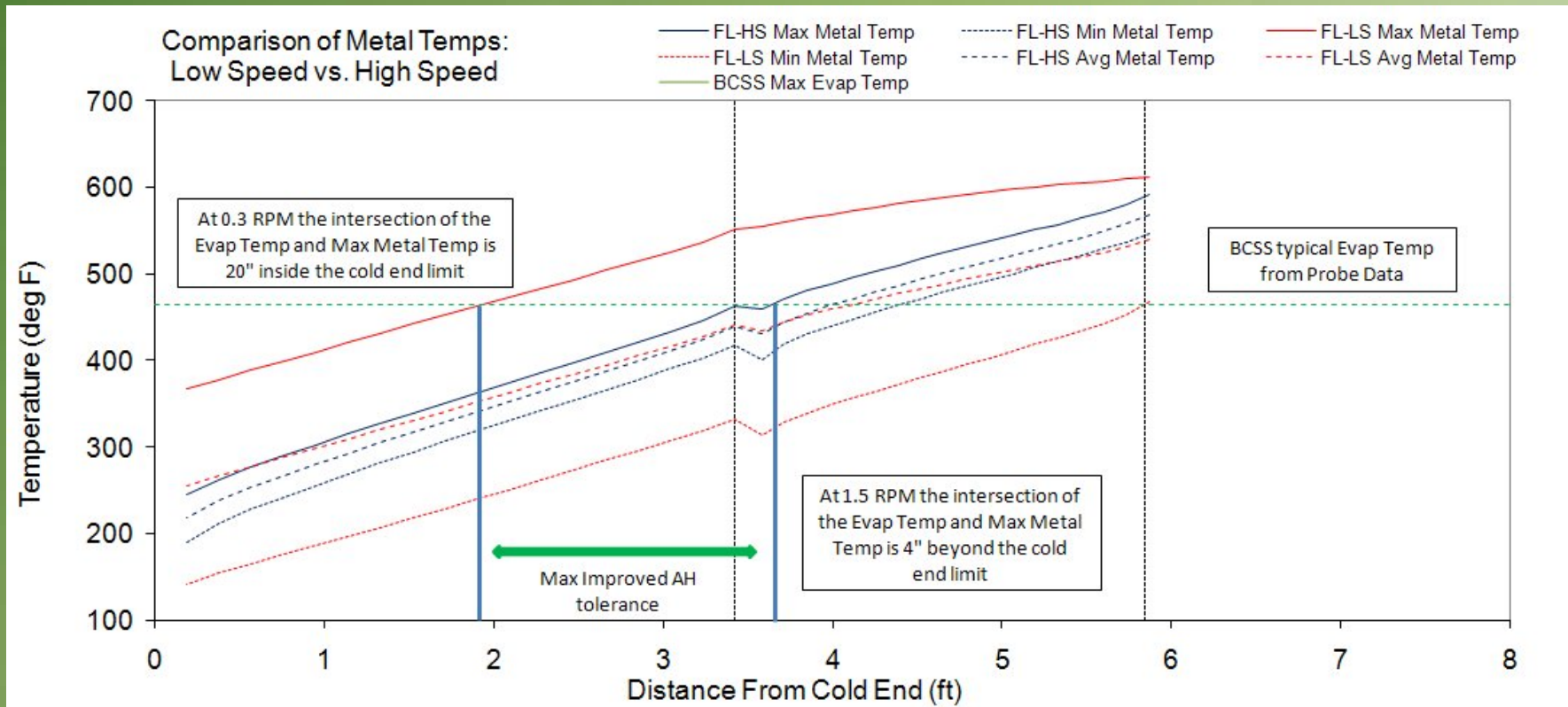


Patent Pending Process

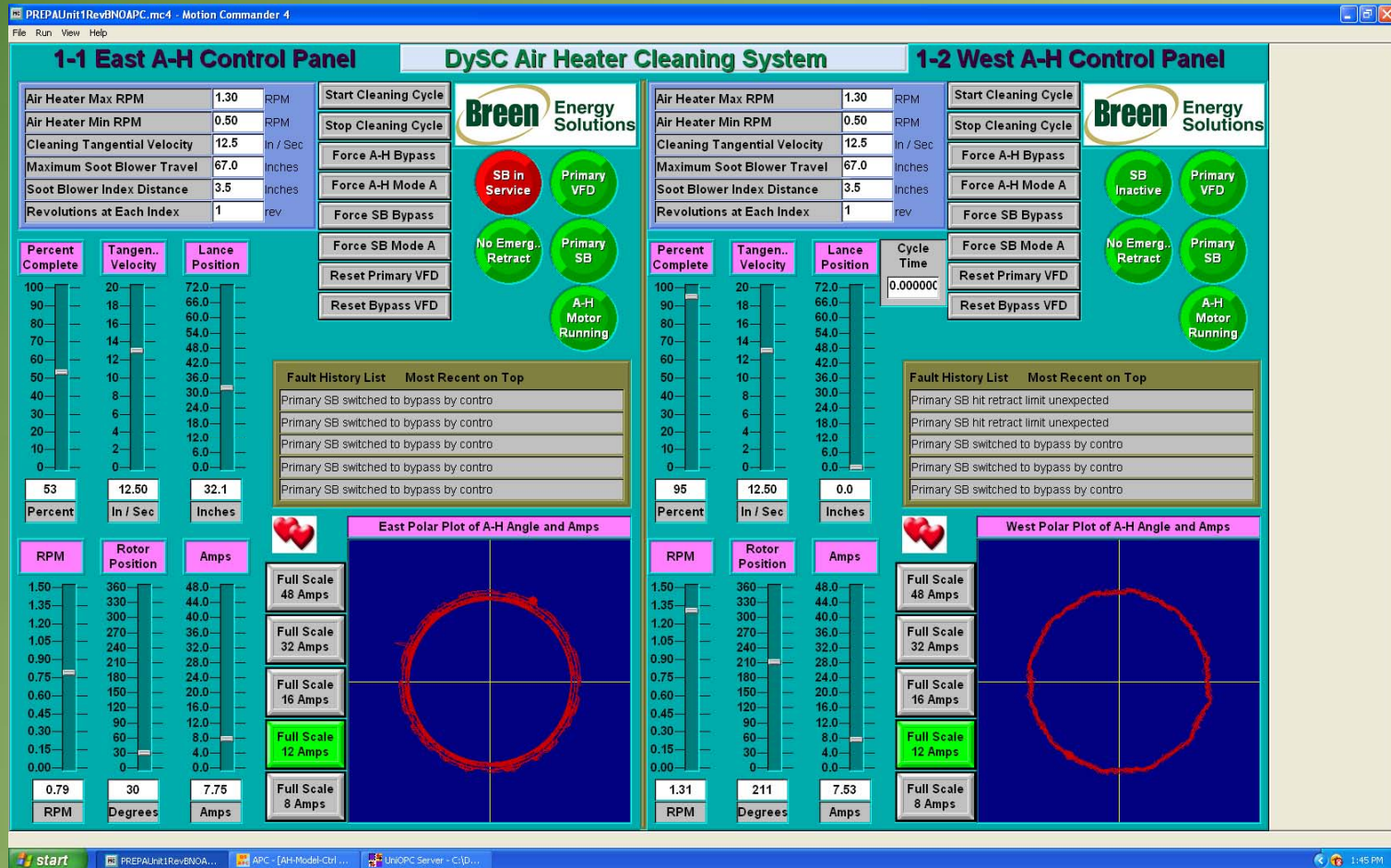
The Breen DySC AH Cleaning Process:

- Positions the Cleaning head at a fixed incremental position
- Controls the AH rotational speed to match the indexed position of the cleaning head
- Assures complete penetration of the cleaning media
- Eliminates AH Pressure Drop problems
- Gives a real time display of individual seal adjustment
- Allows for control of particulate release due to air heater cleaning

Metal Temperature Adjustment



DySC Control



Conclusions

Conclusions

- Integrated Condensables Management can provide significant benefit to SCR operation,
- When constraints exist that prohibit full range of control, Dynamic Sootblowing can provide additional benefit,
- In the end, air heater fouling should cease to be a concern.

Thank You Questions

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